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#### Abstract

The aza-Wittig reactions of benzaldehyde-, acetophenone- and benzophenone 1-[(triphenylphosphoranylidene) amino]ethylidenehydrazones (1) with 2,3-furandiones 6 provide a new route to $4 H, 8 H-1,2,4$ triazolo $[1,5-c][1,3]$ oxazepin-4-ones $\mathbf{1 4}$ or 5,6-dihydro- $7 H, 12 H$-naphtho $[2,1-f][1,2,4]$ triazolo $[1,5-c]$ -[1,3]oxazepin-7-ones $\mathbf{1 7}$ via the thermal reaction of the expected azinoimine vinylogous lactones.


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In recent years, there has been a significant interest in the chemistry of iminophosphoranes because of their utility in the synthesis of a wide variety of nitrogen heterocycles, and many interesting heterocyclization reactions involving functionalized iminophosphoranes have been reviewed [1]. Also, we reported that aza-Wittig reactions of azinoiminophosphoranes $\mathbf{1}$ with aromatic aldehydes [2],

Scheme I

phthalic anhydride and maleic anhydride [3] gave trisubstituted 1,2,4-triazoles 2, $5 H, 7 H$-1,2,4-triazolo[1,5-c]-[1,3]benzoxazepin-7-ones 3 and $6 \mathrm{H}, 8 \mathrm{H}-1,2,4$-triazolo-[1,5-a][1,3]oxazepin-6-ones 4 by thermal rearrangement of intermediate azinoimines, respectively (Scheme I). In addition, Schweizer and co-workers described that the reactions of azine ylides 5 with 2,3-furandiones $\mathbf{6}$ would give the $4 H, 8 H$-pyrazolo[1,5-c][1,3]oxazepin-4-ones $\mathbf{8}$, $4 H, 6 H$-pyrazolo [1,5-c] oxazol-4-ylidines 9 and/or $4 H$ -pyrrolo[1,2-b]pyrazol-4-ones $\mathbf{1 0}$ depending upon R and $\mathrm{R}^{1}$ substituents [4] (Scheme II). With their results in mind, we tried to apply this methodology to the synthesis of $4 H, 8 H-1,2,4$-triazolo[1,5-c][1,3]oxazepin-4-ones 14 and 5,6-dihydro- $7 \mathrm{H}, 12 \mathrm{H}$-naphtho[2,1-f][1,2,4]triazolo[1,5-c]-[1,3]oxazepin-7-ones $\mathbf{1 7}$ by the reaction of known azinoiminophosphoranes 1a-c with furandiones 6a-c (Scheme III and IV).

Scheme II


The aza-Wittig reactions of $\mathbf{1 a - c}$ with furandiones $\mathbf{6 a}$ and $\mathbf{6 b}$ in toluene at reflux temperature led directly to 27$71 \%$ yields of a single product, hitherto unknown $4 \mathrm{H}, 8 \mathrm{H}$ -1,2,4-triazolo[1,5-c][1,3] oxazepin-4-ones 14a-f. Both possible five-membered ring closure products $\mathbf{1 5}$ and $\mathbf{1 6}$ were not produced. When the reactions were carried out at room temperature and monitored by thin-layer chromatography, no reactions occurred. The reactions of iminophosphoranes 1a-c with furandione $\mathbf{6 c}$ were also investigated and reacted smoothly at $90-95^{\circ} \mathrm{C}$ [5] to afford $54-72 \%$ yields of 5,6-dihydro-7H,12H-naphtho[2,1-f][1,2,4]triazolo-[1,5-c][1,3]oxazepin-7-ones 17a-c (Scheme IV).

A proposed mechanism for the formation of 1,2,4-triazoles 14 is shown in Scheme III based on literature [4]. The presumed intermediate azinoimine vinylogous lactones 11 were too unstable to isolate under the reaction condition, so nucleophilic attack by the imine nitrogen on the exocyclic imine bond in $\mathbf{1 1}$ would yield the azomethine imine intermediate $\mathbf{1 2}$ which could open to zwitterionic intermediates 13a-c. Ring closure from 13a would yield the seven-membered 1,2,4-triazole-fused vinylogous lactones 14. Presumably the seven-membered ring closure is a 7-Endo-Trig process and is favored. Both five-membered ring closures to give 15 and $\mathbf{1 6}$ would be a 5-Endo-Trig process and would be disfavored by Baldwin's rule [6].

1a, $R=H$
1b, $\mathrm{R}=\mathrm{Me}$
1c, $\mathrm{R}=\mathrm{Ph}$

6a, $\mathrm{R}^{1}=\mathrm{H}$
6b, $\mathrm{R}^{1}=\mathrm{Me}$

11

12



13 c



16

14c, $\mathrm{R}=\mathrm{Ph}, \mathrm{R}^{1}=\mathrm{H}$
14f, $\mathrm{R}=\mathrm{Ph}, \mathrm{R}^{1}=\mathrm{Me}$

Scheme IV


Structural elucidation of $\mathbf{1 4}$ and $\mathbf{1 7}$ was accomplished on the basis of spectral data. The ${ }^{13} \mathrm{C} \mathrm{nmr}$ spectra showed a characteristic peak at $\delta=175.2-177.5$ for the vinylogous lactone carbonyl carbon, and two triazole carbons resonated in $\delta=152.8$ to 153.5 region and $\delta=158.0$ to 161.6 region. Their infrared spectra showed absorption at 1619$1638 \mathrm{~cm}^{-1}$ for the carbonyl band. Their mass spectra showed molecular ion peaks in all cases.
In conclusion, it has been demonstrated that the azaWittig reactions between azinoiminophosphoranes and furandiones provides a new synthesis of $4 H, 8 H-1,2,4$-tria-zolo[1,5-c][1,3]oxazepin-4-ones 14 and 5,6-dihydro$7 H, 12 H$-naphtho [2,1- $f][1,2,4]$ triazolo $[1,5-c][1,3]-$ oxazepin-7-ones $\mathbf{1 7}$ via the azinoimine vinylogous lactones. This method is useful because we have found no previous reports of the preparation of this ring system.

## EXPERIMENTAL

All reagents and solvents were reagent grade or were purified by standard methods before use and the reactions were routinely carried out under an inert atmosphere. Silica gel 60 (70-230 mesh ASTM) used for column chromatography was supplied by E. Merck. Analytical thin layer chromatography (tlc) was performed on silica gel with fluorescent indicator coated on aluminium sheets. Melting points were taken using an Electrothermal melting point apparatus and are uncorrected. Microanalyses were obtained using a Carlo Erba EA 1180 element analyzer. Mass spectra were obtained using a ThermoQuest Polaris Q mass spectrometer operating at 70 eV . Infrared spectra were recorded on a Nicolet Magna 550 FTIR spectrometer. The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C} \mathrm{nmr}$ spectra were measured on a Gemini 300 spectrometer. All chemical shifts are reported in parts per million ( $\delta$ ) relative to tetramethylsilane in deuteriochloroform solvent.

The benzaldehyde-, acetophenone- and benzophenone 1-[(triphenylphosphoranylidene)amino]ethylidenehydrazones (1a-c) $[2,7]$ and furandiones $6 \mathbf{a - c}$ [8] were prepared following literature procedures.
$4 H, 8 H$-1,2,4-Triazolo[1,5-c] [1,3] oxazepin-4-ones 14a-f and 5,6-Dihydro- $7 H, 12 H$-naphtho[ $2,1-f][1,2,4]$ triazolo[ $1,5-c][1,3]-$ oxazepin-7-ones 17a-c. General Procedure.

To a solution of iminophosphorane $\mathbf{1}$ ( 1.5 mmoles) in 15 ml of toluene was added furandione 6 ( 2.25 mmoles) and this solution was stirred at reflux temperature or at $90-95{ }^{\circ} \mathrm{C}$ [9] for 2 hours. After cooling to room temperature the solvent was removed on a rotary evaporator. The residue was chromatographed on silica

Table 1
$4 H, 8 H-1,2,4$-Triazolo[1,5-c][1,3]oxazepin-4-ones 14a-f and 5,6-Dihydro- $7 H, 12 H$-naphtho[2,1- $f][1,2,4]$ triazolo[1,5-c][1,3]oxazepin-7-ones 17a-c

|  | Reaction <br> Time (h) | Yield <br> (\%) | $\begin{aligned} & \text { MP } \\ & \left({ }^{\circ} \mathrm{C}\right) \end{aligned}$ | Molecular Formula | Analysis (\%) Calcd../Found |  |  | Mass Spectra $\mathrm{m} / \mathrm{z}$ (\%) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  | C | H | N |  |
| 14a | 2 | 27 | 178-180 | $\begin{gathered} \mathrm{C}_{19} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{2} \\ (317.34) \end{gathered}$ | $\begin{aligned} & 71.91 \\ & 71.75 \end{aligned}$ | $\begin{aligned} & 4.76 \\ & 4.61 \end{aligned}$ | $\begin{aligned} & 13.24 \\ & 13.02 \end{aligned}$ | $\begin{aligned} & 317\left(\mathrm{M}^{+}, 7\right), 213(15), 212(100), 173(20), 105(53), \\ & 77(36) \end{aligned}$ |
| 14b | 2 | 47 | 185-187 | $\begin{gathered} \mathrm{C}_{20} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{2} \\ (331.37) \end{gathered}$ | $\begin{aligned} & 72.49 \\ & 72.13 \end{aligned}$ | $\begin{aligned} & 5.17 \\ & 5.03 \end{aligned}$ | $\begin{aligned} & 12.68 \\ & 12.45 \end{aligned}$ | $\begin{aligned} & 331\left(\mathrm{M}^{+}, 18\right), 227(16), 226(100), 221(30), 105(38) \text {, } \\ & 77(21) \end{aligned}$ |
| 14c | 2 | 52 | 211-213 | $\underset{(393.44)}{\mathrm{C}_{25} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{O}_{2}}$ | $\begin{aligned} & 76.32 \\ & 76.08 \end{aligned}$ | $\begin{aligned} & 4.87 \\ & 4.59 \end{aligned}$ | $\begin{aligned} & 10.68 \\ & 10.40 \end{aligned}$ | $\begin{aligned} & 393\left(\mathrm{M}^{+}, 100\right), 392(37), 365(27), 364(28), 336 \text { (57), } \\ & 294(47), 288 \text { (57), } 105(27), 77(28) \end{aligned}$ |
| 14d | 2 | 68 | 207-209 | $\begin{gathered} \mathrm{C}_{20} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{2} \\ (331.37) \end{gathered}$ | $\begin{aligned} & 72.49 \\ & 72.20 \end{aligned}$ | $\begin{aligned} & 5.17 \\ & 4.96 \end{aligned}$ | $\begin{aligned} & 12.68 \\ & 12.41 \end{aligned}$ | 331 ( $\mathrm{M}^{+}, 25$ ), 226 (40), 105 (100), 77 (37) |
| 14e | 2 | 71 | 173-175 | $\begin{gathered} \mathrm{C}_{21} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{O}_{2} \\ (345.39) \end{gathered}$ | $\begin{aligned} & 73.03 \\ & 69.74 \end{aligned}$ | $\begin{aligned} & 5.54 \\ & 5.32 \end{aligned}$ | $\begin{aligned} & 12.17 \\ & 11.82 \end{aligned}$ | 345 ( $\left.\mathrm{M}^{+}, 14\right), 330$ (12), 240 (100), 105 (84), 77 (36) |
| 14 f | 2 | 56 | 174-176 | $\begin{gathered} \mathrm{C}_{26} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{2} \\ (407.46) \end{gathered}$ | $\begin{aligned} & 76.64 \\ & 76.39 \end{aligned}$ | $\begin{aligned} & 5.19 \\ & 5.03 \end{aligned}$ | $\begin{aligned} & 10.31 \\ & 10.58 \end{aligned}$ | $\begin{aligned} & 407\left(\mathrm{M}^{+}, 12\right), 365(34), 330(100), 314(57), 302(62), \\ & 297(53), 285(28), 207(42), 105(69), 77(45) \end{aligned}$ |
| 17a | 2 | 70 | 190-192 | $\begin{gathered} \mathrm{C}_{21} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{2} \\ (343.38) \end{gathered}$ | $\begin{aligned} & 73.45 \\ & 72.23 \end{aligned}$ | $\begin{aligned} & 4.99 \\ & 4.70 \end{aligned}$ | $\begin{aligned} & 12.24 \\ & 12.02 \end{aligned}$ | $\begin{aligned} & 343\left(\mathrm{M}^{+}, 41\right), 326(42), 314(20), 266(69), 233(100) \text {, } \\ & 171(21) \end{aligned}$ |
| 17b | 2 | 54 | 198-200 | $\begin{gathered} \mathrm{C}_{22} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{O}_{2} \\ (357.41) \end{gathered}$ | $\begin{aligned} & 73.93 \\ & 73.70 \end{aligned}$ | $\begin{aligned} & 5.36 \\ & 5.03 \end{aligned}$ | $\begin{aligned} & 11.76 \\ & 11.48 \end{aligned}$ | 357 ( $\mathrm{M}^{+}$, 16), 324 (100), 247 (62), 253 (44), 198 (36) |
| 17e | 2 | 72 | $\begin{aligned} & 209-211 \\ & (419.47) \end{aligned}$ | $\mathrm{C}_{27} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{2}$ | $\begin{aligned} & 77.31 \\ & 77.10 \end{aligned}$ | $\begin{aligned} & 5.05 \\ & 4.87 \end{aligned}$ | $\begin{gathered} 10.02 \\ 9.76 \end{gathered}$ | 419 ( $\left.\mathrm{M}^{+}, 19\right), 342$ (71), 326 (100), 309 (66), 207 (60) |

Table 2
IR and NMR Data of Compounds 14 and 17

IR $v\left(\mathrm{~cm}^{-1}\right)$
(KBr)
14a 1638, 1596, 1568, 1487, 1429, 1335, 1269, 1219, 1114, 1048, 978, 838
14b $1627,1600,1572,1483$, 1448, 1413, 1347, 1258, 1215, 1091, 765
1631, 1603, 1572, 1475, 1444, 1339, 1258, 1122, 1048, 970, 897, 757
14d $1627,1600,1475,1429$, 1332, 1300, 1223, 1153, 1044, 776
14e $1634,1588,1565,1491$, 1444, 1332, 1312, 1277, 1227, 1145, 1095, 1021
14f $1619,1586,1561,1479$, 1452, 1421, 1328, 1258, 1149, 1044, 768 1623, 1603, 1549, 1487, 1436, 1370, 1300, 1246, 1192, 1157, 1102, 733
17b $\quad 1627,1600,1561,1479$, 1421, 1363, 1312, 1161, 1102, 932, 854, 765

17c $1627,1600,1557,1479$, 1448, 1417, 1355, 1308, 1153, 1095, 998, 765

## ${ }^{1} \mathrm{H}$ NMR $\delta(\mathrm{ppm})$ <br> (Deuteriochloroform)

$2.46(\mathrm{~s}, 3 \mathrm{H}), 6.51(\mathrm{~s}, 1 \mathrm{H}), 7.38(\mathrm{~s}, 1 \mathrm{H})$, 7.34-7.58 (m, 8H), 7.77-7.79 (m, 2H)
$2.51(\mathrm{~s}, 3 \mathrm{H}), 2.57(\mathrm{~s}, 3 \mathrm{H}), 6.33(\mathrm{~s}, 1 \mathrm{H})$ 6.77-6.79 (m, 2H), 7.21-7.30 (m, 3H), 7.49-7.58 (m, 3H), 7.86-7.89 (m, 2H) $2.41(\mathrm{~s}, 3 \mathrm{H}), 6.44(\mathrm{~s}, 1 \mathrm{H}), 7.08-7.11$ (m, 4H) 7.34-7.57 (m, 9H), 7.81-7.84 (m, 2H)
$2.07(\mathrm{~s}, 3 \mathrm{H}), 2.41(\mathrm{~s}, 1 \mathrm{H}), 7.13(\mathrm{~s}, 1 \mathrm{H})$ 7.39-7.53 (m, 10H)
$1.76(\mathrm{~s}, 3 \mathrm{H}), 2.39(\mathrm{~s}, 3 \mathrm{H}), 2.56(\mathrm{~s}, 3 \mathrm{H})$ 6.89-6.92 (m, 2H), 7.28-7.33 (m, 3H), 7.46-7.55 (m, 5H)
$1.83(\mathrm{~s}, 3 \mathrm{H}), 2.40(\mathrm{~s}, 3 \mathrm{H}), 7.11-7.14$ (m, 4H), 7.30-7.48 (m, 11H)
$2.46(\mathrm{~s}, 3 \mathrm{H}), 2.71-2.87(\mathrm{~m}, 4 \mathrm{H}), 7.21-$ $7.46(\mathrm{~m}, 8 \mathrm{H}), 7.81-7.83(\mathrm{~m}, 1 \mathrm{H})$
$2.47(\mathrm{~s}, 3 \mathrm{H}), 2.57(\mathrm{~s}, 3 \mathrm{H}), 2.43-2.54(\mathrm{~m}$, $2 \mathrm{H}), 2.63-2.77(\mathrm{~m}, 2 \mathrm{H})$, 6.73-6.75 (m, 2H), 7.15-7.29 (m, 4H), 7.38-7.45 (m, 2H), 8.01-8.03 (m, 1H)
$2.41(\mathrm{~s}, 3 \mathrm{H}), 2.62(\mathrm{~s}, 4 \mathrm{H}), 7.07-7.19(\mathrm{~m}$, 4H), 7.21-7.25 (m, 1H), 7.26-7.44 (m, 8H), 7.87-7.90 (m, 1H)

## ${ }^{13} \mathrm{C}$ NMR $\delta(\mathrm{ppm})$ <br> (Deuteriochloroform)

13.9, 89.4 (d), 108.5, 126.7, 127.4, 128.0, 128.7, $129.3,130.5,132.0,132.9,152.8,161.6,165.0$, 175.2
13.9 (d), 30.5 (d), $96.3,108.4$ (d), 124.4, 127.2, 129.0, 129.1, 129.7, 132.2, 133.5, 139.0, 153.0, 160.7, 162.2, 175.3
14.0, 99.1, 108.6 (d), 127.1, 127.5, 127.8, 128.5, $129.5,130.2,133.4,138.1,153.5,160.4,162.0$, 175.3
13.9, 16.9 (d), 89.1 (d), 117.6, 127.8, 128.4, 128.7, 129.9, 130.1, 131.2, 131.6, 134.5, 152.9, 161.4, 166.0, 176.9
14.0 (d), 16.3 (d), $30.0,95.2,118.1,125.1,128.4$, 128.9, 129.4, 129.5, 130.6, 135.6, 138.6, 153.1, 160.7, 161.8, 177.1
14.1, 16.0 (d), 98.1, 119.6, 127.8, 128.2, 128.5, 129.2, $129.7,130.0,135.6,138.1,153.3,160.6,161.1,177.5$
13.9 (d), 22.2, 27.3, 89.0 (d), 118.1, 125.3, 125.4, $126.9,127.7,128.9,130.1,130.3,131.8,132.5,140.4$, 153.0, 160.3, 161.5, 175.3
13.9 (d), 21.9, 27.2, 30.3 (d), 95.7, 118.0, 124.6, 125.0, 127.1, 127.6, 128.8, 129.5, 130.7, 131.5, 139.1, 140.5, $153.1,158.0,160.6,175.4$
14.1 (d), 21.9, 27.2, 98.6, 118.7, 125.3, 125.4, 127.1,
127.7, 128.3, 129.8, 130.9, 131.5, 138.4, 140.4, 153.5, 158.0, 160.4, 175.5
gel column eluting with hexane-ethyl acetate $3: 1$ to give the product $\mathbf{1 4}$ or $\mathbf{1 7}$. Purification was achieved by crystallization from ethyl ether.
The physical and spectral data of compounds 14 and 17 prepared by this general method are listed in Table 1 and Table 2.
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